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SECTION 1 continued 7. Write net ionic equations that represent the following reactions: a. the ionization of HClO₃ in water HClO₃

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$3(\text{aq}) \text{H}_2\text{O}(\text{l}) \rightarrow \text{H}_3\text{O}^+(\text{aq}) \text{ClO}_3^-(\text{aq})$ b. NH_3 functioning as an Arrhenius base $\text{NH}_3(\text{aq}) \text{H}_2\text{O}(\text{l}) \rightarrow \text{NH}_4^+(\text{aq}) \text{OH}^-(\text{aq})$ 8. a. Explain how strong acid solutions carry an electric current. Strong acids ionize completely in solution. These ions are free to move ...

14 Acids and Bases

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CHAPTER 10 REVIEW States of Matter SECTION 3 SHORT ANSWER Answer the following questions in the space provided.

1. Match description on the right to the correct crystal type on the left. b ionic crystal (a) has mobile electrons in the crystal c

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covalent molecular crystal (b) is hard, brittle, and nonconducting
a metallic crystal (c) typically has the lowest melting point of the
four

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Stoichiometry SECTION 1 SHORT ANSWER Answer the following
questions in the space provided. 1. b The coefficients in a
chemical equation represent the (a) masses in grams of all
reactants and products. ... mc06se_cFMsr_i-vi.qxd Author ...

Modern Chemistry Chapter 9 Worksheet Answers

Modern Chemistry Chapter 15 Acid-Base Titration & pH Section 1
self-ionization of water occurs when two water molecules
produce a hydronium (H_3O^+) and a hydroxide (OH^-) ion 2 H_2O

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$\text{H}_3\text{O}^+ + \text{OH}^-$ - The ionization constant (K_w) of water is: $K_w = [\text{H}_3\text{O}^+][\text{OH}^-] = 1.0 \times 10^{-14}$ M Acidic, Basic, & Neutral IF $[\text{H}_3\text{O}^+] > [\text{OH}^-]$ then solution is acidic. IF $[\text{H}_3\text{O}^+] < [\text{OH}^-]$ then solution is basic. IF $[\text{H}_3\text{O}^+] = [\text{OH}^-]$ then solution is neutral.

Modern Chemistry Chapter 15 Acid-Base Titration & pH

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provided. 1. a When a substance in a closed system undergoes a phase change and the system reaches ...

Modern Chemistry Chapter 10 Review Answers States Of

...

CHAPTER 1 REVIEW Matter and Change MIXED REVIEW SHORT ANSWER Answer the following questions in the space provided.

1. Classify each of the following as a homogeneous or heterogeneous substance. a. sugar d. plastic wrap b. iron filings e. cement sidewalk c. granola bar 2. For each type of investigation, select the most appropriate branch of chemistry from the following

mc06se cFMsr i-vi

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